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Pogil 1. Name 3 strong acids and write their formulas. What makes an acid strong? +hydrochloric acid HCl, nitric acid HNO₃, sulfuric acid H₂SO₄. Strong means 100% ionized, all the H ions have broken off the acid molecule in water. It exists as all ions. 2. Name 4 weak acids and write their formulas.

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2 POGIL™ Activities for High School Biology 5. All the internal structures are suspended (floating) in what substance? 6. One of the bacteria in Model 1 has a tail-like structure. a. What is this structure called? b. What might be the purpose of this structure? c. Based on your answer to the previous question, what might you infer about the ...

Prokaryotic and Eukaryotic Cells

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33_Strong_vs_Weak_Acids-S - Strong versus Weak Acids

What With strong acids this is easy Hydrochloric acid is a strong acid - virtually 100% ionised Each mole of HCl reacts with the water to give 1 mole of hydrogen ions and 1 mole of chloride ions That means that if the concentration of

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HNO₃ – Nitric Acid H₂SO₄ – Sulfuric Acid HClO₄ –

Perchloric Acid Weak acids do not dissociate completely.

Two common (AP questions) weak acids are: HF –

Hydrofluoric Acid HF (aq) + H₂O (l) H₃ Strong Acids,

Weak Acids, Strong Bases, Weak Bases Acids ...

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